

Mathematics & Physics (p. 3 – 10)

I. What Is Physics?

1. Define the term physics.

Physics – branch of science that involves the study of the physical world (energy, matter, motion, electricity, magnetism)

2. List four careers that people who study physics pursue.

- 1. Engineers
- 2. Computer Science
- 3. Teachers
- 4. Construction

Plans after H.S.?

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II. Mathematics & Physics

1. Why are mathematical models needed for physical science experiments & observations?

- To support conclusions. (Neptune discovery)

2. What is an easy way to double-check mathematical solutions?

- Does it make sense (Based on estimation + experience)

III. SI Units

1. Define the term SI units.

SI Units – standards that are universally accepted + understood by scientists throughout the world

2. Match the SI base unit with the correct quantity measured.

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|--------------|-----------------------|------------------|
| 1. <u>E.</u> | - Temperature | A. Kilogram (kg) |
| 2. <u>A.</u> | - Mass | B. Mole (mol) |
| 3. <u>G.</u> | - Luminous Intensity | C. Ampere (A) |
| 4. <u>D.</u> | - Length | D. Meter (m) |
| 5. <u>B.</u> | - Amount Of Substance | E. Kelvin (K) |
| 6. <u>F.</u> | - Time | F. Second (s) |
| 7. <u>C.</u> | - Electric Current | G. Candela (cd) |

Derived Units

Density = $\frac{\text{mass}}{\text{volume}}$

Newton = $\text{kg} \cdot \text{m/s}^2$

Weight = mg

Momentum = $\text{kg} \cdot \text{m/s}$

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3. Conversions between SI units are based on the appropriate power of 10.

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IV. Dimensional Analysis

1. Define the term dimensional analysis.

Dimensional Analysis – method of treating units as algebraic quantities (which can be cancelled)

2. One mile is equal to 1.61 kilometers. What is the speed of a car traveling 55 miles / hour in the units of kilometers / hour?

$$\frac{1 \text{ mile}}{1.61 \text{ km}} = \frac{55 \text{ miles}}{(x) \text{ km}}$$

$$1x = (55)(1.61)$$

$$x = 88.5 \text{ km/hr.}$$

V. Significant Digits**1. Define the term significant digits.**

(Reduces experimental uncertainty)

Significant Digits – valid digits in a measurement**2. List the four rules for determining significant digits. (See page 834.)**Examples:

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1. Nonzero digits are significant.
2. Final zeros after a decimal point are significant.
3. Zeros between two significant digits are significant.
4. Zeros used only as placeholders are not significant.

3. Identify the number of significant digits in each measurement.

	<u>Significant Digits</u>	<u>Significant Digits</u>
1. 1,405 m	4	4
2. 2.50 km	3	3
3. 0.0034 m	2	2
4. 12.007 kg		5
5. 5.8×10^6 kg		2
6. 3.03×10^{-5} mL		3

4. The result of an arithmetic operation can never be more precise than the least-precise measurement.Circle One :TrueFalse**5. How are significant digits determined when adding or subtracting values?**

(1) Perform operation (2) Round off result to least-precise value.

6. How are significant digits determined when multiplying or dividing values?

(1) Perform calculation (2) Round off result to least-precise value.

7. Solve the following equations using the correct number of significant digits.

1. $5.012 \text{ km} + 3.4 \text{ km} + 2.33 \text{ km} = 10.7 \text{ km}$

2. $45 \text{ g} - 8.3 \text{ g} = 37 \text{ g}$

3. $3.40 \text{ cm} \times 7.125 \text{ cm} = 24.2 \text{ cm}$

4. $54 \text{ m} / 6.5 \text{ s} = 8.3 \text{ m/sec}$

VI. Scientific Methods

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1. Match the following terms with the correct definitions.

1. State Problem
2. Gather Information
3. Form Hypothesis
4. Test Hypothesis
5. Analyze Results
6. Draw Conclusion

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| 1. <u>D.</u> | - Scientific Method | A. Educated guess about how variables are related. |
| 2. <u>A.</u> | - Hypothesis | B. Explanation based on observations supporting results.
<i>(Big Bang Theory, Theory of Evolution)</i> |
| 3. <u>E.</u> | - Model | C. Rule of nature that describes a pattern in nature.
<i>(Law of Gravity; Newtonian Laws)</i> |
| 4. <u>C.</u> | - Scientific Law | D. Method of observing, experimenting, and analyzing. |
| 5. <u>B.</u> | - Scientific Theory | E. Concept created as an idea, equation, structure, or system
<i>(Solar System, Geologic Time Scale, Atom, DNA)</i> |